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Antitubercular Nitroimidazoles Revisited: Synthesis and Activity of the Authentic 3-Nitro Isomer of Pretomanid

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ABSTRACT: A published study of structural features associated with the aerobic and anaerobic activities of 4- and 5-nitroimidazoles had found that the 3-nitro isomer of pretomanid, **8**, displayed interesting potencies, including against nitroreductase-mutant *Mycobacterium tuberculosis*. However, recent NMR analyses of two trace byproducts, isolated from early process optimization studies toward a large-scale synthesis of pretomanid, raised structural assignment queries, particularly for **8**, stimulating further investigation. Following our discovery that the reported compound was a 6-nitroimidazooxazole derivative, we developed a *de novo* synthesis of authentic **8** via nitration of the chiral des-nitro imidazooxazine alcohol **26** in trifluoroacetic or acetic anhydride, and verified its identity through an X-ray crystal structure. Unfortunately, **8** displayed no antitubercular activity (MICs >128 μM), whereas the second byproduct (3'-methyl pretomanid) was 8-fold more potent than pretomanid in the aerobic assay. These findings further clarify target specificities for bicyclic nitroimidazoles, which may become important in the event of any future clinical resistance.

Tuberculosis (TB) currently ranks alongside HIV/AIDS as one of the leading causes of global mortality, claiming about 1.4 million lives every year.¹ The spread of multi- and extensively drug-resistant (MDR/XDR) TB is widely regarded as a burgeoning health crisis, placing great strain on limited healthcare resources for only modest treatment outcomes (cure rates of 20-50% following complex therapy for up to 30 months with several more costly and toxic agents).¹⁻³ The recent emergence of programmatically incurable tuberculosis has led to many patients being discharged back into the community, further threatening control efforts.^{4,5} In this context, the conditional approval of two new MDR-TB drugs, delamanid (OPC-67683, **1**; see Figure 1) and bedaquiline (TMC-207, **2**), is a tremendous advance, although access to these and repurposed agents such as linezolid (**3**) remains very limited in many countries.^{4,6} To further address this urgent need, the TB Alliance has been developing shorter acting novel regimens involving the nitroimidazooxazine pretomanid (PA-824, **4**).^{7,8} Initial phase III clinical results for the combination of **2**, **3**, and **4** (Nix-TB) in XDR-TB patients look highly encouraging, with 29 of 31 patients who completed the treatment and 6 month follow-up being cured.⁹ Another regimen that combines **2** and **4** with pyrazinamide (**5**) and moxifloxacin (**6**) is also demonstrating excellent bactericidal efficacy against both drug-sensitive and MDR-TB.¹⁰

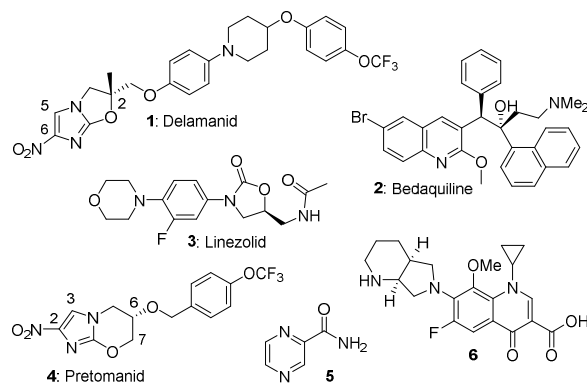


Figure 1. Drugs included in novel regimens for MDR/XDR-TB

The mechanism of action of **4** involves bioreductive activation by a deazaflavin (F420) dependent nitroreductase (Ddn), leading to several metabolites arising from reduction of the imidazole ring at C-3, including a des-nitro derivative.¹¹ Formation of the latter correlates with the release of nitric oxide, a crucial element in the *anaerobic* bactericidal activity of **4**, which is considered important for treatment shortening.¹² In contrast, the aerobic killing effects of **4** are attributed to the inhibition of cell wall mycolic acid biosynthesis.¹³ The 5-

nitroimidazole metronidazole also has some antitubercular activity under hypoxic conditions (*vide infra*) but a clinical study in MDR-TB patients showed that it was too neurotoxic for long term use.¹⁴

In the past decade, several SAR studies of the nitroimidazooxazine class have been reported by ourselves and others,^{15,16} seeking more effective second-generation analogues of **4**. While the predominant focus has centered on optimising the aryl side chain (e.g., clinical candidate TBA-354¹⁷), a few studies examined alternative linkages at C-6¹⁸ or more fundamental modifications to the nitroheterobicyclic “warhead”.^{19,20} Here, replacement of the nitroimidazole portion by nitrotriazole or nitropyrazole abolished activity, whereas the 8-oxygen could be exchanged for sulfur or nitrogen, and substitution at C-7 was tolerated.²¹ Notably, Kim et al.²² also described the synthesis and biological evaluation of the 3-nitro isomer of **4**, which was recorded as being “only slightly less active” (5- to 10-fold) than **4** itself, whereas its precursor alcohol derivative was 16- to 31-fold less active than **4**. Furthermore, the anaerobic killing of mutant *Mycobacterium tuberculosis* (*M. tb*) by this isomeric compound was suggested to imply a different biological target. This result could be of particular significance in the event of future clinical resistance to **4**. Nevertheless, in the related nitroimidazooxazole class (*cf.* **1**), we discovered that relocating the nitro group from C6 to C5 reduced the aerobic MIC against *M. tb* by at least 2-3 orders of magnitude.²³

During initial attempts to optimize a large-scale synthesis of **4**, two trace byproducts were isolated (<0.5%), and one of these was postulated to be the 3-nitro isomer, although its NMR data did not match those provided by Kim et al.²² To establish its identity, we sought to make this 3-nitroimidazooxazine analogue of **4** via an unambiguous chiral synthesis. Concurrently, we also targeted the second compound, thought to be a 3'-methyl derivative of **4**. We now report the intriguing findings from this study, including some preliminary biological assessments.

To prepare the 3-nitro isomer of **4**, we first investigated the method of Kim²² (Scheme 1A). Reaction of 2-chloro-4-nitroimidazole (**10**) with the TBS ether derivative of *R*-glycidol (**11**) gave two products in a ratio of ~3:1, with the major one being the expected 4-nitroimidazole derivative **12**, as described. However, after low temperature crystallization of the more polar oily minor product (reportedly the 5-nitroimidazole isomer of **12**, i.e. **13**), we found that its ¹H NMR spectrum in CDCl₃ contained a D₂O-exchangeable hydroxyl proton resonance at δ_{H} 1.83 ppm, which appeared as a sharp “dd” ($J = 6.7, 5.1$ Hz), and coupled to the two proton resonance of a methylene group in a COSY experiment. Moreover, a NOESY experiment revealed a strong NOE effect between the imidazole proton resonance and the proton resonances from the directly attached N-methylene. HMBC correlations between H-5 and this NCH₂ carbon and between the NCH₂ protons and both C-2 and C-5 (Figure 2) confirmed that the correct structure of this minor product was the 4-nitroimidazole **16**. This result is in good accordance with the known tendency of a TBS group to migrate under basic conditions, as employed here.²⁴

A critical consequence of this structural assignment error is that subsequent THP protection of the primary hydroxyl of **16** and treatment with TBAF (as reported) would induce cyclization to a 6-nitroimidazooxazole (**18**), rather than the 3-nitroimidazooxazine **15** (Scheme 1A). Therefore, following

THP deprotection and installation of the 4-(trifluoromethoxy)benzyl ether, Kim et al. finally obtained **20** (instead of **8**), equivalent to the 2*S* enantiomer of compound **10** in our nitroimidazooxazole paper,²³ as verified by their identical ¹H and ¹³C NMR data. We note that while our compound had been prepared by an independent and unequivocal method, we have rigorously confirmed its structure by 2D NMR, including a NOESY experiment, where an NOE effect was observed between the imidazole proton resonance (H-5) and resonances from the adjacent methylene protons at the 3-position in the oxazole ring. This revised structure (**20**) for the compound claimed by Kim et al. as **8** is also consistent with the much smaller than expected optical rotation value obtained ($[\alpha]_{\text{D}}^{20} +7.4$ *cf.* -44.7 for **4** in MeOH²⁵) as well as the moderate antitubercular potency recorded (*vide infra*), as *S* enantiomers are known to be an order of magnitude less effective than *R* forms in the 2*H* nitroimidazooxazole class.²³

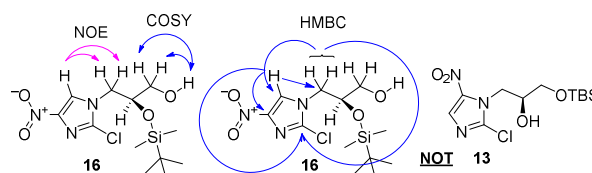
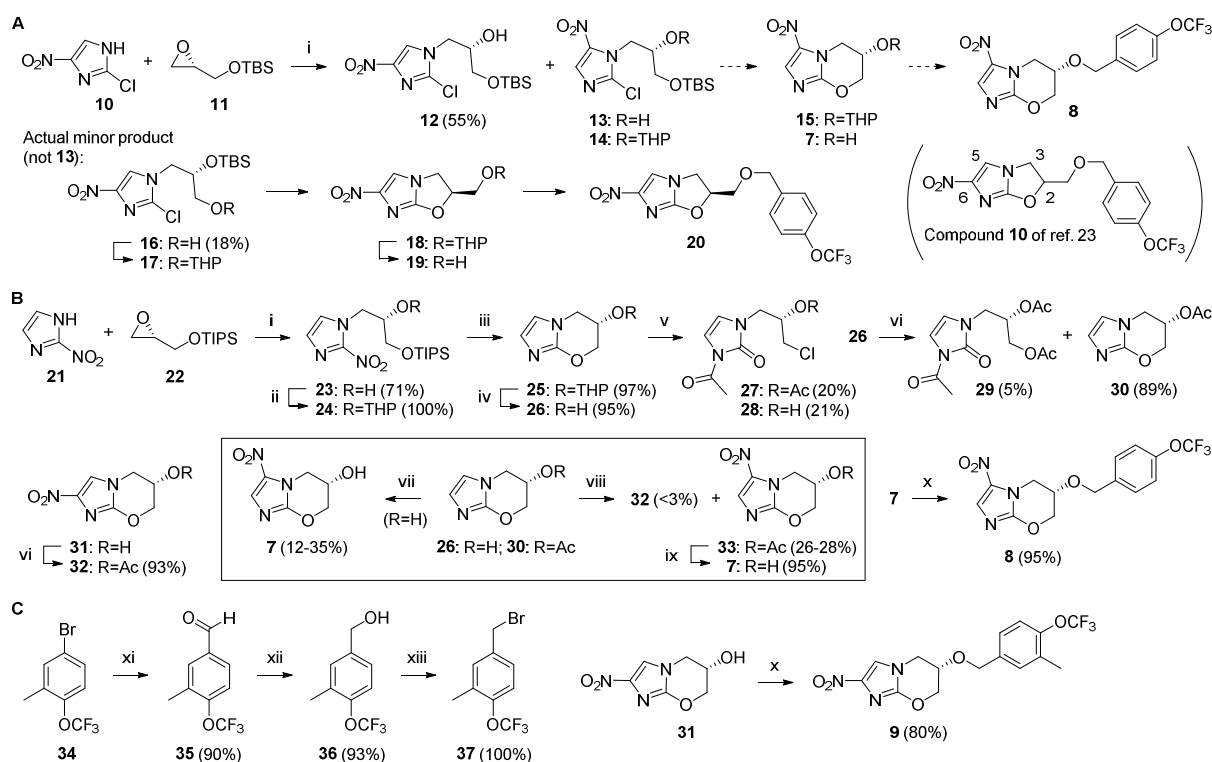


Figure 2. 2D NMR evidence for structure **16** over the reported **13**

In an alternative approach to the 3-nitro isomer of **4**, we explored the novel nitration of known²² imidazooxazine alcohol **26** (Scheme 1B). This strategy was based on previous success in our laboratory with the nitration of a related 5,6,7,8-tetrahydroimidazo[1,2-*a*]pyridin-6-ol in acetic anhydride,¹⁹ as well as literature precedence for the regioselective C-5 nitration of both 1-alkyl-2-(alkylthio)imidazoles²⁶ and 2,3-dihydroimidazo[2,1-*b*][1,3]thiazole.²⁷ During our initial synthesis of **26**, we again encountered a problem with migration of the TBS group when glycidyl ether **11** was reacted with 2-nitroimidazole (**21**).²² Fortunately, this was easily overcome by switching the protecting group to the more stable TIPS (**22**).^{24,25} We also made improvements to the subsequent THP protection and deprotection steps by adopting methods from Orita et al.,²⁵ enabling the synthesis of alcohol **26** in 65% yield over 4 steps. Surprisingly, attempted preparation of its acetate ester (**30**) using acetyl chloride in pyridine gave only ring opened N-acetyl imidazol-2-one products, **27** and **28** (having a distinctive ¹H NMR methyl resonance²⁸ at δ_{H} 2.64 ppm); this was largely circumvented by employing acetic anhydride.

Several reagent systems were then explored for the nitration of **26** or **30** (Table 1). Neither heating in aqueous nitric acid nor treatment with acetyl nitrate in acetic anhydride gave the desired product, while a reaction of **30** with nitronium tetrafluoroborate was low-yielding and accompanied by minor side products. Overall, the best conditions were 70% nitric acid in trifluoroacetic anhydride²⁹ or a mixture of concentrated nitric and sulfuric acids in acetic anhydride, which enabled yields of ~25-30% of the 3-nitro derivatives, **7** and **33**, respectively. Intriguingly, only one nitro isomer was formed in trifluoroacetic anhydride, but this reaction proved to be more capricious than the acetic anhydride alternative, where the isomer ratio was ~9:1 in favour of nitration at C-3 over C-2 (for confirmation, known²⁵ 2-nitro acetate ester **32** was prepared from alcohol **31**³⁰). Isomers **32** and **33** were separable by careful column chromatography and cleavage of acetate **33** to alcohol **7** was cleanly achieved through the use of a mild base (NaHCO₃).

Scheme 1. Synthesis of the target compounds 7-9, highlighting a key nitration step^a

^aReagents and conditions: (i) K₂CO₃, EtOH, 70 °C, 20 h (for **12/16**), or 75-82 °C, 23 h (for **23**); (ii) 3,4-dihydro-2H-pyran, PPTS, toluene, 20 °C, 5 d; (iii) TBAF, THF, 124 °C, 24 h (sealed tube); (iv) concd HCl (1.1 equiv), MeOH, 20 °C, 15 h; (v) AcCl, pyridine, 0-20 °C, 4 h; (vi) Ac₂O, pyridine, 0-20 °C, 7-19 h; (vii) TFAA, -5 to 0 °C, 15 min, then concd HNO₃, -50 to 20 °C, 3 h, then ice, NaHCO₃; (viii) Ac₂O, 0-20 °C, 5-80 min, then concd HNO₃, concd H₂SO₄, -50 to 0 °C, 1.5-2.5 h, then ice, NaHCO₃; (ix) NaHCO₃, aq MeOH, 20 °C, 5 h; (x) 4-OCF₃BnBr or **37**, NaH, DMF, 0-20 °C, 2.5-3.3 h; (xi) *n*BuLi, THF, -78 °C, 1 h, then DMF, -78 to 20 °C, 1.5 h, and then aq citric acid; (xii) NaBH₄, MeOH, 0-20 °C, 1.5 h; (xiii) HBr, AcOH, 20 °C, 13 h.

Table 1. Summary of nitration methods explored for the synthesis of alcohol **7** or acetate **33**

Compd	Solvent	Reagents (equiv)	Temp range	Time/temp	Products (% yield)
26	H ₂ O	3M HNO ₃ (9)	80 to 97 °C	3 h/97 °C	-
26	H ₂ O	50% HNO ₃ (>100)	90 °C	2 h/90 °C	-
26 ^a	Ac ₂ O	70% HNO ₃ (3.1) ^b	-15 to 20 °C	16 h/20 °C	-
26 ^a	Ac ₂ O	96% H ₂ SO ₄ (1.7), 100% HNO ₃ (3.0)	-55 to 0 °C	60 min/0 °C	32 (<3), 33 (20) ^c
26 ^a	Ac ₂ O	96% H ₂ SO ₄ (1.7), 70% HNO ₃ (3.0)	-50 to 0 °C	75 min/0 °C	32 (<3), 33 (26) ^c
26 ^a	Ac ₂ O	96% H ₂ SO ₄ (2.5), KNO ₃ (1.7)	-35 to -30 °C	20 min/-30 °C	32 (<2), 33 (14) ^c
26 ^a	TFAA	70% HNO ₃ (2.5-2.8)	-50 to 20 °C	2-3 h/20 °C	7 (12-35)
30	CH ₃ CN	NO ₂ BF ₄ (1.5)	-48 to 20 °C	90 min/20 °C	30 (15), 32 (1), 33 (14) ^c
30	Ac ₂ O	96% H ₂ SO ₄ (1.9), 70% HNO ₃ (3.2)	-50 to 0 °C	30 min/0 °C	32 (<3), 33 (28) ^c

^aAcetylated or trifluoroacetylated in situ, prior to nitration. ^bPreformed acetyl nitrate. ^cYields after chromatography and crystallization.

Importantly, all of the successful nitration methods gave a complete retention of stereochemistry (100% ee by chiral HPLC analysis, using the analogously synthesized racemic form of **7** as a reference standard). This indicated that the in situ formed trifluoroacetate or acetate esters were sufficiently stable to protect the chiral alcohol from potential racemisation, e.g., through the formation of a chiral nitrate ester and subsequent hydrolysis, as this can induce inversion of configuration.³¹ Alkylation of pure **7** with 4-(trifluoromethoxy)benzyl bromide (NaH/DMF) then gave the desired target **8** in excellent yield (95%) and, gratifyingly, this proved to be identical

(by NMR, mp, HPLC, and optical rotation) to the first by-product derived from optimization studies for a large-scale process route to **4**. As expected from findings in the nitroimidazoazooxazole class, the rotation value for **8** {[α]_D²⁴ -150.7 (c 1.002, CHCl₃)} was indeed much larger than the one recorded for **4**.²⁵ Conclusive structural proof was gained through a single crystal X-ray structure (Figure 3); of note, the benzyloxy side chain adopted a pronounced pseudoaxial conformation at C6, the same as that observed in the previously reported crystal structure of **4**.²¹

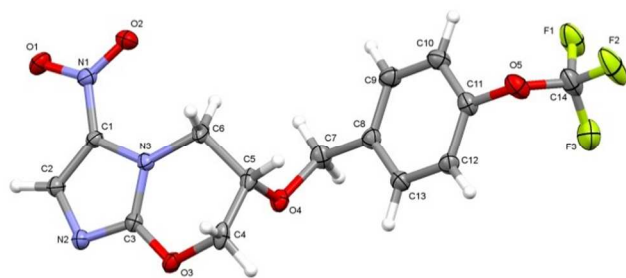


Figure 3. X-ray crystal structure of compound **8**

Preparation of the second byproduct from the synthesis of **4** was straightforward (Scheme 1C). Lithiation of bromotoluene **34** and quenching with DMF gave the required aldehyde **35**, which was easily reduced (NaBH_4) and brominated to give **37**. Alkylation of chiral alcohol **31**³⁰ with bromide **37** then provided the expected compound **9**, which was also identical to that derived from the non-optimized process chemistry route for **4**. This byproduct (<0.2%) was postulated to arise from traces of **37** in the bulk commercial 4-(trifluoromethoxy)benzyl bromide. Full experimental procedures and characterization data for all compounds have been provided in the Supporting Information.

The comparative effects of **7-9** and several reference drugs against *M. tb* (strain H37Rv) were assessed in two *in vitro* assays, MABA³² and LORA,³³ conducted under aerobic and hypoxic conditions, respectively. The latter assay employed bacteria preadapted to low oxygen conditions and represented a targeted screen for the identification of agents with better sterilizing ability against non-replicating persistent bacteria. Recorded minimum inhibitory concentrations (MICs) corresponding to growth inhibitions of $\geq 90\%$ (Table 2) were mean values obtained from replicate measurements (\pm standard deviation). Compounds **7-9** were also evaluated for cytotoxicity against mammalian cells (VERO) in a 72 h assay³² and found to be non-toxic ($\text{IC}_{50} > 128 \mu\text{M}$).

Table 2. *In vitro* activities of **7-9** versus other TB drugs

Compd	MIC ^a (μM)		IC ₅₀ ^b (μM)
	MABA	LORA	
2	0.070 \pm 0.018	0.11 \pm 0.03	>10
3	2.9 \pm 1.2	2.8 \pm 0.3	
4 ^c	0.50 \pm 0.30	2.6 \pm 1.4	>128
6	0.42 \pm 0.13	>128	
7	>128	>128	>128
8	>128	>128	>128
9	0.063 \pm 0.003	1.0 \pm 0.1	>128
MET	>512	79 \pm 40	
RMP	0.049 \pm 0.027	0.64 \pm 0.35	>100
INH	0.34 \pm 0.18	>128	

^aMinimum inhibitory concentration against *M. tb*, determined under aerobic (MABA)³² or hypoxic (LORA)³³ conditions. Each value is the mean of ≥ 2 independent determinations (**7-9** were tested 3 times). The controls were metronidazole (MET), rifampicin (RMP), and isoniazid (INH). ^bIC₅₀ values for cytotoxicity toward VERO cells. ^cMIC data from ref. 15.

As expected from our studies in the nitroimidazooxazole class,²³ 3-nitro compounds **7** and **8** were completely inactive in both *M. tb* assays (MICs >128 μM ; Table 2). For **8**, this indicates a >256-fold loss in activity in comparison to its 2-nitro isomer, **4**. Conversely, the 3'-methyl derivative of **4** (**9**) was 7- to 8-fold more active than both **4** and moxifloxacin (**6**) in MABA (similar to rifampicin and bedaquiline **2**) and ~ 3 -fold better than both **4** and linezolid (**3**) in LORA (comparable to rifampicin). These results for **9** were in accordance with the findings reported by Cherian et al.¹⁶ for the 3'-methoxy derivative of **4**, which was 5-fold superior to **4** in the aerobic assay and 2-fold more effective than **4** under hypoxic conditions.

In their original investigation, Kim et al.²² reported an aerobic MIC₉₉ value of 4-8 μM and weak anaerobic activity (MIC₉₀ 31 μM) for the compound they believed to be **8** (shown here to be the 2*H* nitroimidazooxazole **20**). Intriguingly, the same compound also yielded an anaerobic MIC₉₀ value of 62.5-125 μM against *M. tb* having mutations in the nitroreductase Ddn, suggesting the participation of a different biological target. While delamanid (**1**) was shown to be primarily triggered by Ddn, several Ddn homologues have been implicated in the cellular activation of simple 2*H* nitroimidazooxazoles (e.g., 2-Et, 2-Ph).³⁴ However, in the case of **20**, the lack of significant potency against this Ddn mutant under aerobic conditions (MIC₉₉ >100 μM) renders this explanation unsatisfactory. Overall, the results of the current investigation support findings from previous studies of resistance to **4** that the activation of nitroimidazooxazines relies exclusively on Ddn.³⁵ In this class, relocation of the nitro group from C-2 to C-3 destroys all antitubercular activity, implying that, like the *R*-enantiomer of **4**,³⁴ **8** is not a substrate for Ddn or its homologues and does not release nitric oxide.

In summary, we set out to establish the identities of two novel byproducts from optimization studies around a manufacturing route to **4** through a combination of 2D NMR analysis and *de novo* chiral synthesis. In the case of 3-nitro isomer **8**, this entailed the development of an innovative nitration route, following our discovery of a critical structural assignment error in the published method, and we obtained an X-ray crystal structure of this compound for final confirmation. Preliminary *in vitro* assessments indicated that whereas the 3'-methyl derivative of **4** (**9**) was markedly more effective than **4**, both **8** and its alcohol precursor **7** were completely inactive, overturning previous misconceptions regarding their aerobic and anaerobic activities and the suggested involvement of another target facilitating their activity against Ddn mutant *M. tb*. These results provide further clarity of fundamental SAR for pretomanid and of the structural features of antitubercular nitroimidazoles that are more likely to overcome any future clinical resistance to **4**.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Further background, scheme for preparation of racemic **7**, experimental procedures and characterizations for compounds, combustion analytical data, packing diagram for X-ray structure of **8**, crystallographic data, NMR spectra for key compounds, chiral HPLC trace for **7** (PDF)

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

TB, tuberculosis; *M. tb*, *Mycobacterium tuberculosis*; MDR, multidrug-resistant; XDR, extensively drug-resistant; Ddn, deazaflavin-dependent nitroreductase

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