# **Uncompetitive** inhibitor

**Uncompetitive inhibition**, also known as **anti-competitive inhibition**, takes place when an <u>enzyme</u> <u>inhibitor</u> binds only to the complex formed between the <u>enzyme</u> and the <u>substrate</u> (the E-S complex). Uncompetitive inhibition typically occurs in reactions with two or more substrates or products.<sup>[1]</sup>

While uncompetitive inhibition requires that an enzyme-substrate complex must be formed, <u>non-competitive inhibition</u> can occur with or without the substrate present.



General representation of uncompetitive inhibition

Uncompetitive inhibition is distinguished from

competitive inhibition by two observations: first uncompetitive inhibition cannot be reversed by increasing [S] and second, as shown, the <u>Lineweaver–Burk plot</u> yields parallel rather than intersecting lines. This behavior is found in the inhibition of <u>acetylcholinesterase</u> by tertiary amines ( $R_3N$ ). Such compounds bind to the enzyme in its various forms, but the acyl-intermediate-amine complex cannot break down into enzyme plus product.<sup>[2]</sup>

# Contents

#### Mechanism

**Mathematical definition** 

Implications and uses in biological systems

Involvement in cancer mechanisms

Importance in cell and organelle membranes

Presence in the cerebellar granule layer

References

### Mechanism

As inhibitor binds, the amount of ES complex is reduced. This reduction in the effective concentration of the ES complex can be explained by the fact that having the inhibitor bound to the ES complex essentially converts it to ESI complex, which is considered a separate complex altogether. This reduction in ES complex decreases the maximum enzyme activity ( $V_{max}$ ), as it takes longer for the substrate or product to leave the active site. The reduction in  $K_m$  - the substrate concentration at which the enzyme can operate at half of its maximal velocity, often used to approximate an enzyme's affinity for a substrate - can also be linked back to the decrease in ES complex. Le Chatelier's principle opposes this decrease and attempts to make up for the loss of ES, so more free enzyme is converted to the ES form, and the amount of ES increases overall. An increase in ES generally indicates that the enzyme has a high degree of affinity for its substrate.  $K_m$  decreases as affinity for a substrate increases, though it is not a perfect predictor of affinity since it accounts for other factors as well; regardless, this increase in affinity will be accompanied by a decrease in  $K_m$ .

In general, uncompetitive inhibition works best when substrate concentration is high. An uncompetitive inhibitor need not resemble the substrate of the reaction it is inhibiting. At no concentration of substrate will the activity of the enzyme be higher when an uncompetitive inhibitor is present, but at low concentrations of substrate the enzyme activity difference will be negligible.<sup>[4]</sup>

### **Mathematical definition**

The Lineweaver–Burk equation states that:

$$rac{1}{v} = rac{K_m}{V_{max}[S]} + rac{1}{V_{ ext{max}}}$$

Where *v* is the initial reaction velocity,  $K_m$  is the Michaelis–Menten constant,  $V_{\text{max}}$  is the maximum reaction velocity, and [*S*] is the concentration of the substrate.<sup>[5]</sup>

The Lineweaver–Burk plot for an uncompetitive inhibitor produces a line parallel to the original enzyme-substrate plot, but with a higher <u>y</u>-

<u>intercept</u>, due to the presence of an inhibition term  $\frac{[I]}{K}$ .

$$rac{1}{v} = rac{K_m}{V_{max}[S]} + rac{1+rac{[I]}{K_i}}{V_{max}}$$

Where [*I*] is the concentration of the inhibitor and  $K_i$  is an inhibition constant characteristic of the inhibitor.<sup>[6][7]</sup>

The Michaelis-Menten equation is altered to:

$$V_0 = rac{V_{max}[S]}{K_m + lpha'[S]}$$

where

$$lpha' = 1 + rac{[I]}{K_I'}$$
 and  $K_I' = rac{[ES][I]}{[ESI]}$ 

As described by the equation above, at high concentrations of substrate,  $V_0$  approaches  $V_{max}/\alpha'$ . Thus, an uncompetitive inhibitor lowers the measured  $V_{max}$ . Apparent  $K_m$  also decreases, because [S] required to reach one-half  $V_{max}$  decreases by the factor  $\alpha'$ .<sup>[8]</sup> It is important to note that  $V_{max}$  and  $K_m$  decrease at the same rate as a result of the inhibitor.<sup>[4]</sup> This is apparent when viewing a Lineweaver-Burk plot of uncompetitive enzyme inhibition: the ratio between V and  $K_m$  remains the same with or without an inhibitor present.

## Implications and uses in biological systems

The unique traits of uncompetitive inhibition lead to a variety of implications for the inhibition's effects within biological and biochemical systems. Uncompetitive inhibition is present within biological systems in a number of ways. In fact, it often becomes clear that the traits of inhibition specific to uncompetitive inhibitors, such as



their tendency to act at their best at high concentrations of substrate, are essential to some important bodily functions operating properly.<sup>[9]</sup>

### Involvement in cancer mechanisms

Uncompetitive mechanisms are involved with certain types of cancer. Human <u>alkaline phosphatases</u> such as CGAP have been found to be over-expressed in certain types of cancers, and those phosphotases often operate via uncompetitive inhibition. It has also been found that a number of the genes that code for human alkaline phosphatases (TSAPs) are inhibited uncompetitively by amino acids such as <u>leucine</u> and <u>phenylalanine</u>.<sup>[10]</sup> Studies of the involved amino acid residues have been undertaken in attempts to regulate alkaline phosphatase activity and learn more about said activity's relevance to cancer.<sup>[11]</sup>

Additionally, uncompetitive inhibition works alongside TP53 to help repress the activity of cancer cells and prevent tumorigenesis in certain forms of the illness, as it inhibits G6PD (glucose-6-phosphate dehydrogenase, an enzyme primarily involved in certain metabolic pathways). One of the side roles G6PD is responsible for is helping to regulate is the control of reactive oxygen levels, as reactive oxygen species must be kept at appropriate levels to allow cells to survive. When G6PD's substrate concentration is high, uncompetitive inhibition of the enzyme becomes far more effective.<sup>[12]</sup> As substrate concentration increases, the amount of ES complex increases as well, and with more ES complex to bind, uncompetitive inhibitors become far more active. This inhibition works such that the higher the concentration of substrate is in the system initially, the harder it is to reach the maximum velocity of the reaction. At low initial substrate concentrations, increasing the concentration of substrate is sometimes enough to entirely or even fully restore the enzyme's function, but as soon as initial concentration increases past a certain point, reaching the maximal enzyme velocity is all but impossible.<sup>[9]</sup> This extreme sensitivity to substrate concentration within the cancer mechanism implicates uncompetitive inhibition rather than mixed inhibition, which displays similar traits but is often less sensitive to substrate concentration due to some inhibitor binding to free enzymes regardless of the substrate's presence.<sup>[9]</sup> As such, the extreme strength of uncompetitive inhibitors at high substrate concentrations and the overall sensitivity to substrate amount indicates that only uncompetitive inhibition can make this type of process possible.

#### Importance in cell and organelle membranes

Although this form of inhibition is present in various diseases within biological systems, it does not necessarily only relate to pathologies. It can be involved in typical bodily functions. For example, active sites capable of uncompetitive inhibition appear to be present in membranes, as removing lipids from cell membranes and making active sites more accessible through conformational changes has been shown to invoke elements resembling the effects of uncompetitive inhibition (i.e. both  $K_M$  and  $V_{Max}$  decrease). In mitochondrial membrane lipids specifically, removing lipids decreases the alpha-helix content in mitochondria and leads to changes in ATPase resembling uncompetitive inhibition.<sup>[13]</sup>

This presence of uncompetitive enzymes in membranes has also been supported in a number of other studies. A type of protein called an <u>Arf protein</u> involved in regulating membrane activity was being studied, and it was found that an inhibitor called BFA trapped one of Arf's intermediates via uncompetitive inhibition. This made it clear that this type of inhibition exists within various types of cells and organelles as opposed to just in pathological cells. In fact, BFA was found to relate to the activity of the Golgi apparatus and its role in regulating movement across the cell membrane. [14]

### Presence in the cerebellar granule layer



Inhibited NMDA Receptor. Substrate is bound and the active site is blocked by the (red) inhibitor.

Uncompetitive inhibition can play roles in various other parts of the body as well. It is part of the mechanism by which <u>NMDA (N-methyl-D-aspartate)</u> <u>glutamate receptors</u> are inhibited in the brain, for example. Specifically, this type of inhibition impacts the granule cells that make up a layer of the cerebellum. These cells have the aforementioned NMDA receptors, and the activity of said receptors typically increases as ethanol is consumed. This often leads to withdrawal symptoms if said ethanol is removed. Various uncompetitive blockers act as antagonists at the receptors and modify the process, with one example being an inhibitor called memorting <sup>[15]</sup> In fact in similar cases (involving of



NMDA Inhibitor Memantine

memantine.<sup>[15]</sup> In fact, in similar cases (involving over-expression of NMDA, though not necessarily via ethanol), it has been shown that uncompetitive inhibition

helps in nullifying the over-expression due to its particular properties. Since uncompetitive inhibitors block high concentrations of substrates very efficiently, their traits alongside the innate characteristics of the receptors themselves lead to very effective blocking of NMDA channels when they are excessively open due to massive amounts of NMDA agonists.<sup>[16]</sup>

### References

- 1. Vladimir L (2004). *Comprehensive Enzyme Kinetics*. Kluwer Academic Publishers. <u>ISBN 978-0306467127</u>. <u>OCLC 517776240</u> (https://www.worldcat.org/oclc/517776240).
- Mathews CK, van Holde KE, Appling DR, Anthony-Cahill SJ (February 26, 2012). *Biochemistry* (4 ed.). Pearson. <u>ISBN</u> <u>978-0138004644</u>.
- 3. Ahern K, Rajagopal I, Tan T (2017). *Biochemistry Free For All Version 1.2*. North Carolina: Creative Commons. pp. 367–368.
- 4. Athel C (2014). *Principles of Enzyme Kinetics*. Elsevier Science. <u>ISBN</u> <u>978-1483164670</u>. OCLC <u>897021733 (https://www.worldcat.org/oclc/897021733)</u>.
- Cleland WW (February 1963). "The kinetics of enzyme-catalyzed reactions with two or more substrates or products. II. Inhibition: nomenclature and theory". *Biochimica et Biophysica Acta* (*BBA*) - Specialized Section on Enzymological Subjects. **67**: 173–87. <u>doi:10.1016/0926-6569(63)90226-8</u> (https://doi.org/10.1016%2F0926-6569%2863%2990226-8). <u>PMID</u> 14021668 (https://pubmed.ncbi.nlm.nih.gov/14021668).
- 6. Rhodes D. "Enzyme Kinetics Single Substrate, Uncompetitive Inhibition, Lineweaver-Burk Plot" (http://www.hort.purdue.edu/rhodcv/hort640c/enzkin1/enzkin1.htm). Purdue University. Retrieved 31 August 2013.
- Cornish-Bowden A (January 1974). "A simple graphical method for determining the inhibition constants of mixed, uncompetitive and non-competitive inhibitors" (https://www.ncbi.nlm.nih.go v/pmc/articles/PMC1166095). The Biochemical Journal. 137 (1): 143–4. doi:10.1042/bj1370143 (https://doi.org/10.1042%2Fbj1370143). PMC 1166095 (https://www.nc bi.nlm.nih.gov/pmc/articles/PMC1166095). PMID 4206907 (https://pubmed.ncbi.nlm.nih.gov/42 06907).
- 8. Nelson DL, Cox MM (November 21, 2012). *Lehninger Principles of Biochemistry* (6 ed.). W.H. Freeman. <u>ISBN</u> 978-1429234146.
- Nahorski SR, Ragan CI, Challiss RA (August 1991). "Lithium and the phosphoinositide cycle: an example of uncompetitive inhibition and its pharmacological consequences". *Trends in Pharmacological Sciences.* **12** (8): 297–303. doi:10.1016/0165-6147(91)90581-C (https://doi.or g/10.1016%2F0165-6147%2891%2990581-C). PMID 1658998 (https://pubmed.ncbi.nlm.nih.go v/1658998).

- Millán JL (July 1992). "Alkaline phosphatase as a reporter of cancerous transformation". *Clinica Chimica Acta; International Journal of Clinical Chemistry*. **209** (1–2): 123–9. <u>doi:10.1016/0009-8981(92)90343-O (https://doi.org/10.1016%2F0009-8981%2892%2990343-O)</u>. <u>PMID 1395034 (https://pubmed.ncbi.nlm.nih.gov/1395034)</u>.
- Millán JL, Fishman WH (1995). "Biology of human alkaline phosphatases with special reference to cancer". *Critical Reviews in Clinical Laboratory Sciences*. **32** (1): 1–39. doi:10.3109/10408369509084680 (https://doi.org/10.3109%2F10408369509084680). PMID 7748466 (https://pubmed.ncbi.nlm.nih.gov/7748466).
- Nyce JW (November 2018). "Detection of a novel, primate-specific 'kill switch' tumor suppression mechanism that may fundamentally control cancer risk in humans: an unexpected twist in the basic biology of TP53" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC6106910). *Endocrine-Related Cancer.* 25 (11): R497–R517. doi:10.1530/ERC-18-0241 (https://doi.org/10. 1530%2FERC-18-0241). PMC 6106910 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC61069 10). PMID 29941676 (https://pubmed.ncbi.nlm.nih.gov/29941676).
- Lenaz G, Curatola G, Mazzanti L, Parenti-Castelli G (November 1978). "Biophysical studies on agents affecting the state of membrane lipids: biochemical and pharmacological implications". *Molecular and Cellular Biochemistry*. 22 (1): 3–32. doi:10.1007/bf00241467 (https://doi.org/10.1 007%2Fbf00241467). PMID 154058 (https://pubmed.ncbi.nlm.nih.gov/154058).
- Zeghouf M, Guibert B, Zeeh JC, Cherfils J (December 2005). "Arf, Sec7 and Brefeldin A: a model towards the therapeutic inhibition of guanine nucleotide-exchange factors". *Biochemical Society Transactions*. 33 (Pt 6): 1265–8. doi:10.1042/BST20051265 (https://doi.org/10.1042%2 FBST20051265). PMID 16246094 (https://pubmed.ncbi.nlm.nih.gov/16246094).
- Tabakoff B, Hoffman PL (March 1993). "Ethanol, sedative hypnotics, and glutamate receptor function in brain and cultured cells". *Behavior Genetics*. 23 (2): 231–6. doi:10.1007/BF01067428 (https://doi.org/10.1007%2FBF01067428). PMID 8390239 (https://pu bmed.ncbi.nlm.nih.gov/8390239).
- Nakamura T, Lipton SA (January 2008). "Emerging roles of S-nitrosylation in protein misfolding and neurodegenerative diseases". *Antioxidants & Redox Signaling*. **10** (1): 87–101. <u>doi:10.1089/ars.2007.1858 (https://doi.org/10.1089%2Fars.2007.1858)</u>. <u>PMID</u> <u>17961071 (http s://pubmed.ncbi.nlm.nih.gov/17961071)</u>.

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