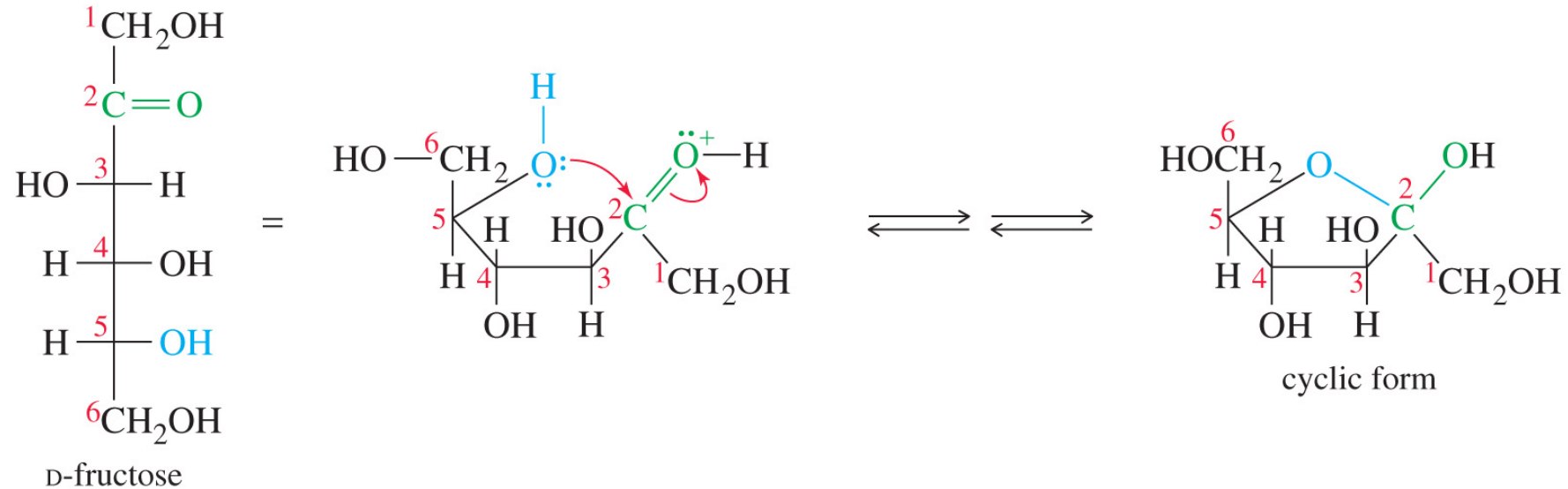


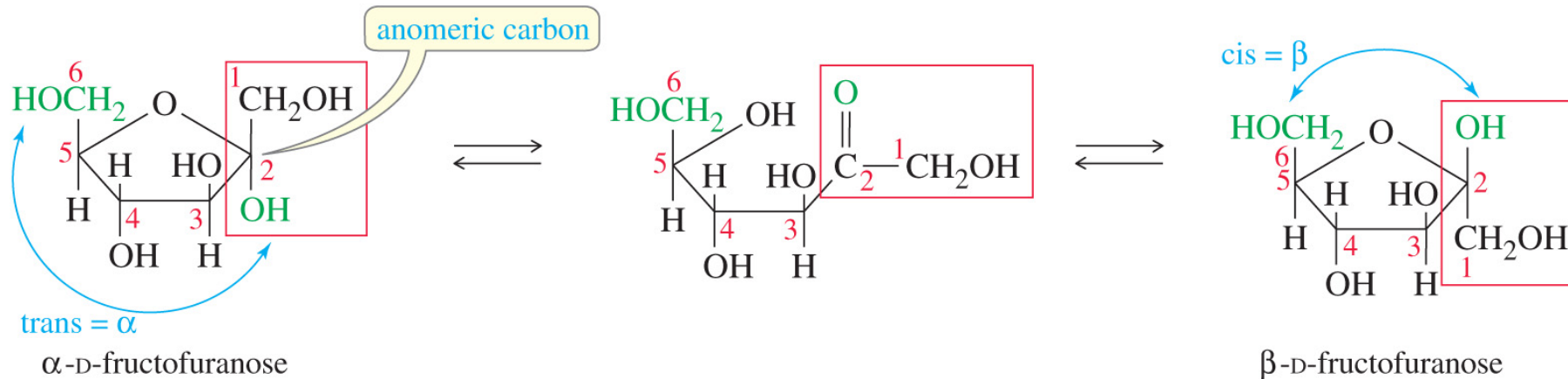
Cyclic Structure for Fructose



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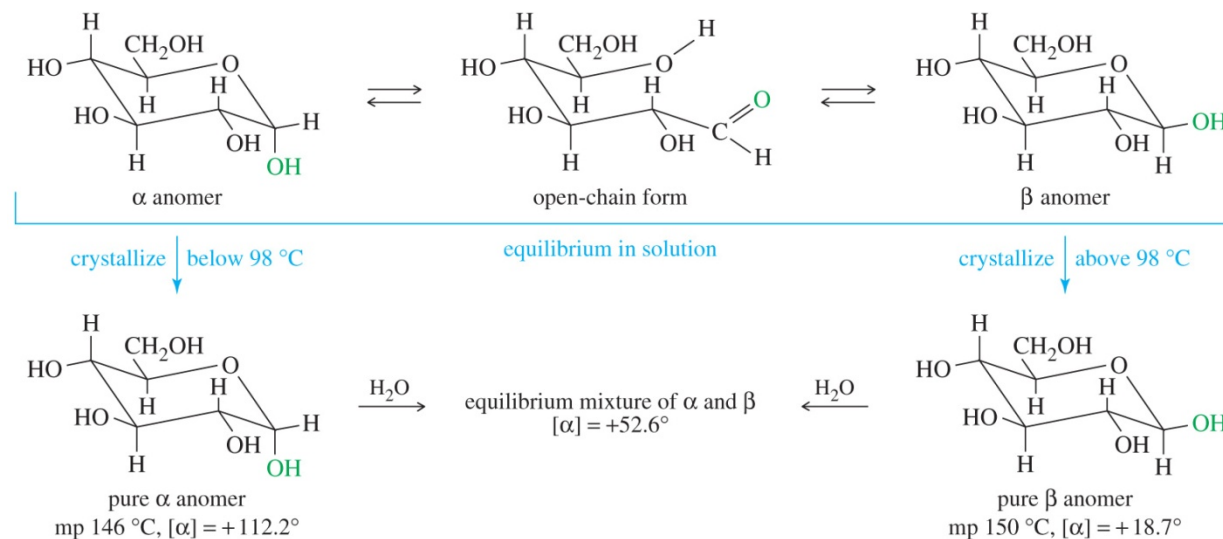
- Cyclic hemiacetal formed by reaction of C=O at C2 with —OH at C5.
- Since five-membered rings are not puckered as much as six-membered rings, they are usually depicted as flat Haworth projections.

Anomers of Fructose



- The α anomer of fructose has the anomeric —OH group down, trans to the terminal — CH_2OH group.
- The β anomer has the anomeric —OH group up, cis to the terminal — CH_2OH .

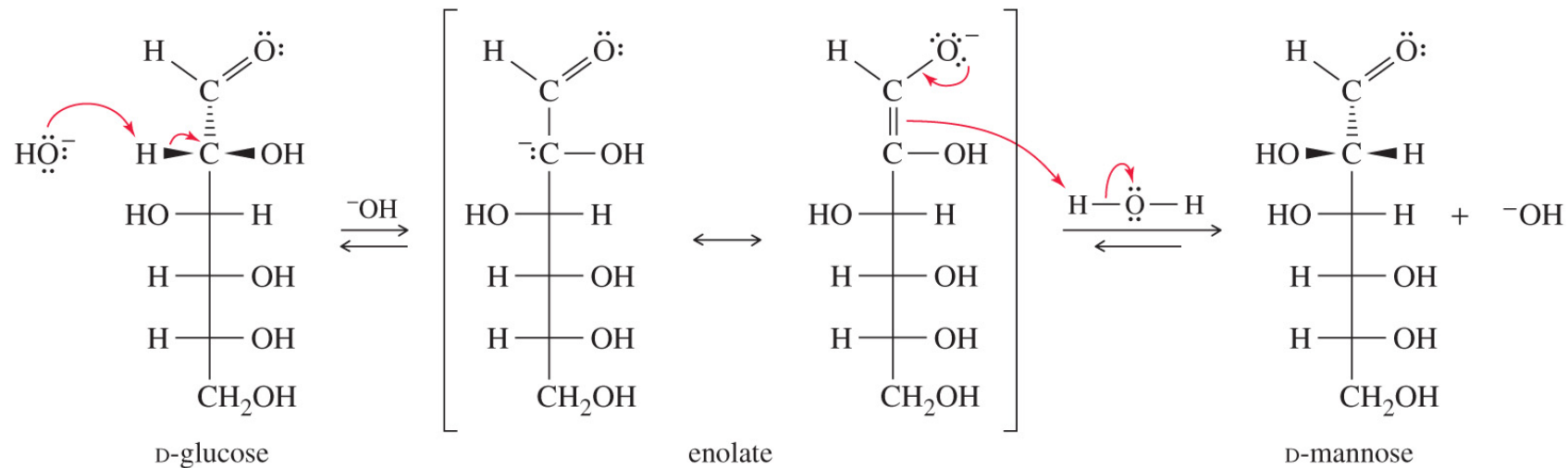
Mutarotation



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- An aqueous solution of D-glucose contains an equilibrium mixture of α -D-glucopyranose, β -D-glucopyranose, and the intermediate open-chain form.
- Crystallization below 98°C gives the α anomer, and crystallization above 98°C gives the β anomer.

Base-Catalyzed Epimerization of Glucose

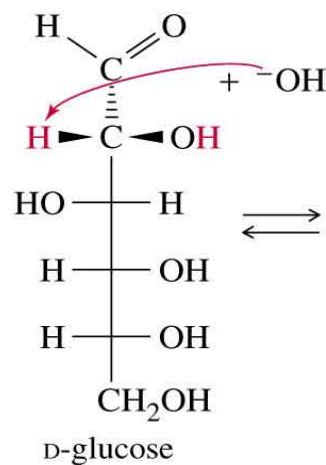


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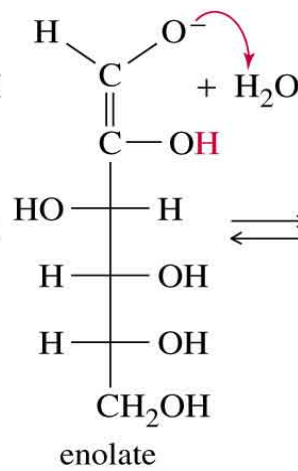
- Under basic conditions, stereochemistry is lost at the carbon atom next to the carbonyl group.
- The enolate intermediate is not chiral, so reprotonation can produce either stereoisomer.
- Because a mixture of epimers results, this stereochemical change is called **epimerization**.

Enediol Rearrangement

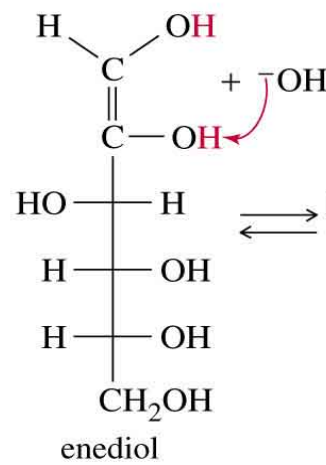
Step 1: Remove the α proton



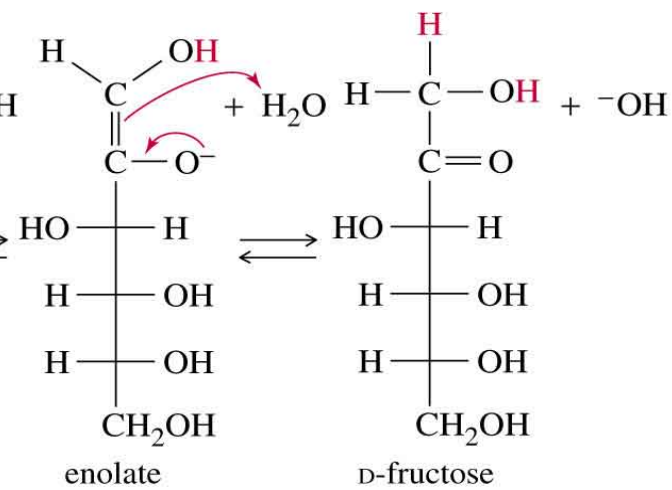
Step 2: Reprotonate on O



Step 3: Deprotonate the O on C2



Step 4: Reprotonate on C1



- In base, the position of the carbonyl can shift.
- Chemists use acidic or neutral solutions of sugars to prevent this rearrangement.