

DISTRIBUTION FUNCTIONS

①

We work in the canonical ensemble
 in the absence of external forces (b)

$P(\bar{R}_1, \dots, \bar{R}_N) \equiv$ probability to have particle 1 in \bar{R}_1
~~probability~~ particle 2 in \bar{R}_2
 (probability density) particle \vdots N in \bar{R}_N

IN THE CANONICAL ENSEMBLE

$$P(\bar{R}_1, \dots, \bar{R}_N) = \frac{1}{Z} e^{-\beta U(\bar{R}_1, \dots, \bar{R}_N)}$$

↑ normalizing factor

Now we define [one-particle probability]

$P^{(1)}(R_1)$ probability (density) to have particle 1 in \bar{R}_1
IRRESPECTIVE of the positions of the
 other particles.

We should simply integrate over the positions of the other particles

$$P^{(1)}(R_1) = \frac{1}{Z} \int dr_2 \dots dr_N e^{-\beta H(R_1, r_2, r_3 \dots r_N)}$$

$P^{(2)}(R_1, R_2)$ probability (density) to have particle 1 in \bar{R}_1
 and particle 2 in \bar{R}_2 , irrespective
 of the positions of the other
 particles

$$P^{(2)}(R_1, R_2) = \frac{1}{Z} \int dr_3 \dots dr_N e^{-\beta H(R_1, R_2, r_3 \dots r_N)}$$

DEFINITION:

$$g^{(2)}(R_1, R_2) = \frac{P^{(2)}(R_1, R_2)}{P^{(1)}(R_1)P^{(1)}(R_2)} \quad \text{PAIR DISTRIBUTION FUNCTION}$$

Such a quantity gives the correlations among the positions of the particles

SOME LIMITING CASES:

(a) ideal gas

particles do not interact.

The probability to have two particles in \bar{R}_1 and \bar{R}_2 is simply the product of the one-particle probabilities.

$$P^{(2)}(R_1, R_2) = P^{(1)}(R_1)P^{(1)}(R_2) \Rightarrow g^{(2)}(R_1, R_2) = 1$$

(b) large distances

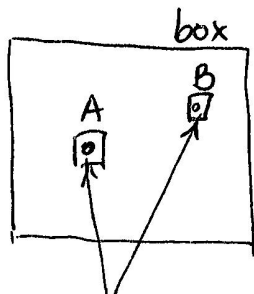
We only consider SHORT-RANGE INTERACTIONS

If the distance between the two particles is much larger than the interaction range the two particles do not influence each other

Again, there is factorization

$$g^{(2)}(R_1, R_2) \approx 1 \quad \text{for } |R_1 - R_2| \text{ large}$$

The absence of external forces implies that the system is HOMOGENEOUS (apart from the region close to the boundaries)



The probability to be in A and B is the same



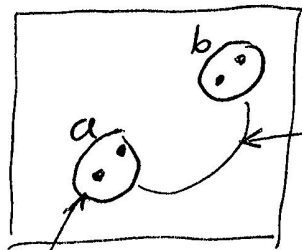
$p^{(1)}(R)$ does not depend on R (boundary effects are negligible)



$$p^{(1)}(R) = \frac{1}{V} \quad \text{because} \quad \int_{\text{box}} d\vec{R} p^{(1)}(R) = 1$$

It follows that $g^{(2)}(R_1, R_2)$ is only a function of $\bar{R}_1 - \bar{R}_2$

$$g^{(2)}(R_1 - R_2) = V^2 p^{(2)}(\bar{R}_1, \bar{R}_2)$$



The probability of the 2-particle configuration in a is the same as that in region b



$p^{(2)}(R_1, R_2)$ depends only on $\bar{R}_1 - \bar{R}_2$

the two configurations are obtained by using a rigid translation

Finally

$$\begin{aligned}
 p^{(2)}(\bar{R}_1, \bar{R}_2) &= \frac{1}{Z} \int dr_3 \dots dr_N e^{-\beta U(r_1, r_2, r_3 \dots r_N)} \\
 &= \frac{1}{Z} \int dr_1 \dots dr_N \delta(r_1 - \bar{R}_1) \delta(r_2 - \bar{R}_2) e^{-\beta U(r_1 \dots r_N)} \\
 &= \langle \delta(\bar{r}_1 - \bar{R}_1) \delta(\bar{r}_2 - \bar{R}_2) \rangle
 \end{aligned}$$

So that

$$g^{(2)}(R_1 - R_2) = V^2 \langle \delta(\bar{r}_1 - R_1) \delta(\bar{r}_2 - R_2) \rangle$$

Now change variable

$$\bar{R}_{12} = \bar{R}_1 - \bar{R}_2 \quad \text{relative position}$$

$$\bar{R}_{CM} = \frac{1}{2} (R_1 + R_2) \quad \text{center-of-mass position}$$

obtaining

$$g^{(2)}(R_{12}) = V^2 \langle \delta(\bar{r}_1 - R_{CM} - \frac{R_{12}}{2}) \delta(r_2 - R_{CM} + \frac{R_{12}}{2}) \rangle$$

Now, we average over \bar{R}_{CM}

$$g^{(2)}(R_{12}) = \frac{1}{V} \int d^3 \bar{R}_{CM} g^{(2)}(R_{12}) =$$

$$= V \int d^3 \bar{R}_{CM} \langle \delta(\bar{r}_1 - R_{CM} - \frac{R_{12}}{2}) \delta(r_2 - R_{CM} + \frac{R_{12}}{2}) \rangle$$

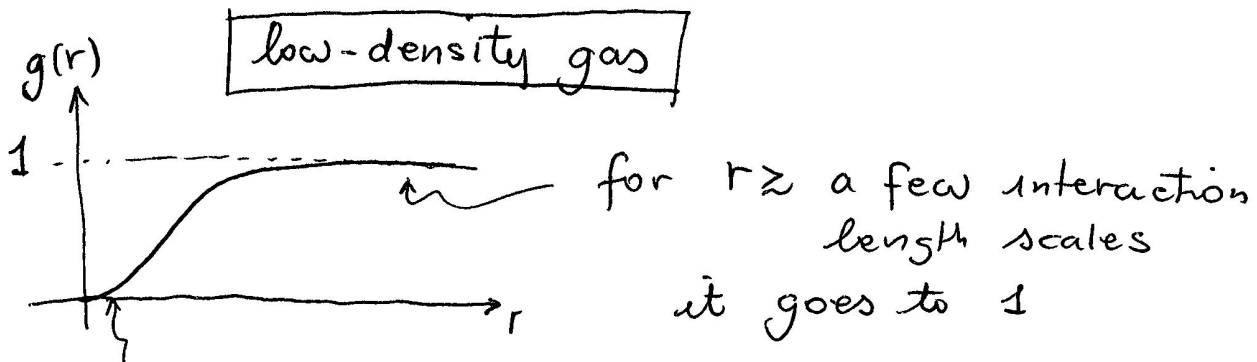
↓
the δ -function implies $R_{CM} = r_1 - \frac{R_{12}}{2}$

$$= V \langle \delta(r_2 - r_1 + \frac{R_{12}}{2} + \frac{R_{12}}{2}) \rangle =$$

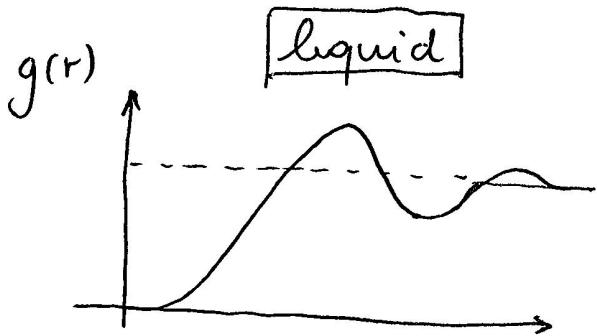
$$= V \langle \delta(r_2 - r_1 + R_{12}) \rangle = V \langle \delta(\bar{r}_1 - \bar{r}_2 - \bar{R}_{12}) \rangle$$

Note that homogeneity also implies that $g^{(2)}(\mathbf{R})$ is only a function of $|\mathbf{R}|$, the distance between the two particles.

Typical graphs of $g^{(2)}(r)$
 \uparrow
 scalar variable: distance



for r small it vanishes: the probability to have two particles in the same position is zero



The presence of oscillations indicates the presence of short-range order: There are preferred distances for the molecules.